

FESENKO, D., inzh.

Increase the efficiency of the production enterprises of road  
construction units. Avt.dor. 27 no.12:22~23 D 164.

(MIRA 18:2)

FESSENKO, E.G.

The crystal structure of zoisite. P. G. Fesenko, I. M. Runganova, and N. V. Belov. Dokl. Akad. Nauk S.S.R. 102, 275-8 (1955).—The structure of zoisite ( $\text{Ca}_2\text{Al}_2\text{Si}_4\text{O}_10(\text{OH})$ ) was determined from x-ray diagrams of colorless crystals in which the Fe content is negligible. The following values were obtained:  $a = 10.20$ ,  $b = 5.50$ , and  $c = 10.14$  Å; the space group is  $D_{10}\bar{h}$ .  $Pmn\bar{m}$ , and the unit cell contains 4 mols. These values agree with earlier data ( $C.I. 49.67\%$ ). The interatomic distances are given as: Si—O, 1.58–1.70 Å; in the Si tetrahedra O—O, 2.63–2.82 Å; Al—O, in the basic columns, 1.03–1.04 Å; and in the octahedra, 1.79–2.03 Å. The valence angle Si—O—Si in the diortho group is  $161^\circ$ .  
I. Rovtar Leach

5

RDN  
RM

~~FESHEKO, Fridrikh Dmitriyevich; LUDSKOV, B.P., red.; BABICHEVA, V.V., tekhn.~~  
~~red.~~

[Organization of accounting machine centers and offices in commerce]  
Organizatsiia mashinoschetnykh stantsii i biuro v torgovle. Moskva,  
Gos. izd-vo torg. lit-ry, 1957. 120 p. (MIRA 11:7)  
(Machine accounting)

- FESENKO, F. D  
MAKAROV, A.; FESENKO, F. (Khar'kov).

A book about machine accounting ("Organization of the machine accounting of trade operations" by V.I. Isakov. Reviewed by A. Makarov, F. Fesenko).  
Sov. terg. no.12:45-47 D '57.  
(MIRA 10:12)

(Machine accounting)  
(Isakov, V.I.)

SITALO, M.V.; FESENKO, G.A.; SITALO, V.M.

Improvement of technological flow sheets and automation of operations  
in the coal preparation plant. Koks i khim. no.5:14-18 '63.  
(MIRA 16:5)

1. Zaporozhskiy koksokhimicheskiy zavod (for M.Sitalo, Fesenko).
2. Zaporozhskiy filial Instituta avtomatiki Gosplana UkrSSR (for  
V.Sitalo).

(Zaporozh'ye---Coal preparation plants)

SITALO, V.M.; KUDRYASHOV, A.N.; NESTEROV, V.V.; FESENKO, G.A.

Automation of the pyramid-shaped thickener. Koks i khim. no.10:  
13-17 '63. (MIRA 16:11)

1. Zaporozhskiy filial Instituta avtomatiki Gosplana UkrSSR (for  
Sitalo, Kudryashov). 2. Institut avtomatiki Gosplana UkrSSR (for  
Nesterov). 3. Zaporozhskiy koksokhimicheskiy zavod (for Fesenko).

ACCESSION NR: AT4045010

S/0000/64/000/000/0160/0164

AUTHOR: Chizh, V. A.; Rudoy, V. S.; Rulla, N. V.; Chekmarev, I. A.; Fesenko, G. M.;  
Nesterova, N. N.

TITLE: Quality control of high-alloy austenitic steel ingots by the method of  
Gamma-defectoscopy

SOURCE: Soveshchaniye po probleme Izpol'zovaniye atomnoy energii. Kiev, 1961.  
Radiatsionnaya avtomatika, izotopy\* i yadernyye izlucheniya v naуke i tekhnike  
(Radiation automation control systems; Isotopes, and nuclear radiation in science  
and technology); doklady\* soveshchaniya. Kiev, izd-vo AN UkrSSR, 1964, 160-164

TOPIC TAGS: steel ingot, steel casting, steel forging, high alloy steel, austenitic  
steel, steel ingot structure, steel ingot defect, ingot defect detection, Gamma  
defectoscopy

ABSTRACT: Air bubbles, porosities and blow holes are common defects in ingots of  
high-alloy austenitic steel. Because of the low plasticity of such steel at high  
temperatures, these defects lead to cracks and porosity and even to complete rup-  
ture of the ingot during forging and rolling. In order to facilitate the detection  
of such defects in steel ingots, the authors tested the method of  $\gamma$ -defectoscopy  
and compared the results with the behavior of the ingots during forging. Eleven  
Cord 1/2

ACCESSION NR: AT4045010

Ingots (80 x 270 mm) were examined by transillumination with  $\gamma$ -rays from Co-60, revealing deep bubbles and porosities in nearly all cases. During subsequent forging to a diameter of 40-43 mm (3-5 forgings with a 350-kg pneumatic hammer at 1150-1180°C), the 2 ingots with the deepest bubbles broke completely, and several others showed defective behavior, thus confirming the effectiveness and accuracy of the  $\gamma$ -defectoscopic technique. Finally, sections (3 cylindrical and 5 conical) were cut from the ingots and the compressibility was tested. The maximal critical compression (10%) was obtained in a section which was free of defects, showing that the plasticity is decreased by both bubbles and porosity. The authors conclude that quality control by this method will permit establishment of maximal permissible limits for defects in steel ingots, which is of particular importance in the case of ingots intended for pipe manufacture. Orig. art. has: 2 figures and 1 table.

ASSOCIATION: none

SUBMITTED: 07Jan64

ENCL: 00

SUB CODE: MM, IE

NO REF Sov: 000

OTHER: 000

Card 2/2

ACCESSION NR: AR4018343

8/0137/64/000/001/r121/r121

SOURCE: RZh. Metallurgiya, Abs. 11784

AUTHOR: Alferova, N. S.; Devterov, V. M.; Fesenko, G. M.

TITLE: Heat-treatment of EI852 steel in the production of pipe

CITED SOURCE: Sb. Proiz-vo trub. Vy\*p. 9, M., Metallurgizdat, 1963, 106-113

TOPIC TAGS: Steel processing, pipe-rolling, EI852 steel; heat treatment, structural conversion, steel hardness

TRANSLATION: For the purpose of determining the optimum schedule of heat-treating EI852 steel composed (in%) of C 0.10-0.15; Si 1.4-2.1; Cr 12-14, Mo 1.2-2.0, Mn less than 0.6, Ni less than 0.5, which is used extensively in the production of pipe, structural conversions and changes in the hardness of hot and cold-rolled pipe of this steel with hardening at 800-1,200° were studied. The critical point of EI852 steel, beginning with which, during heating, there takes place a partial conversion of ferrite-carbide mixture into austenite, equal to approximately 925 degrees. To obtain satisfactory plasticity in longitudinal and lateral directions, to remove percussion marks of hot-rolled Me and to form the structure of grainy.

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ACCESSION NR: AR4018343

perlite in the sections of products of conversion of austenite, for hot-rolled pipe of EI852 steel, it is recommended to use double heat-treatment according to the following schedule: heating from 1,000-1,050 degrees, with subsequent drawing at 800-820 degrees. It is recommended to use drawing at 800-830 degrees with a time interval of more than one hour as a form of intermediate heat treatment for cold-rolled pipe.

SUB CODE: IR, MM ENCL: 00

Card 2/2

FESENKO, I., starshiy nauchnyy sotrudnik

Observe crop rotations. Zashch. rast. ot vred. i bol. 10  
no.1:17 '65. (MIRA 18:3)

1. Moldavskaya optytnaya stantsiya Vsesoyuznogo nauchno-issledo-  
vatel'skogo instituta maslichnykh i efiromaslichnykh kul'tur.

FESENKO, I.A.

DOLGOPOLOV, N.N.; BEZHUKOV, P.L., redaktor; BUSHINSKIY, G.I., redaktor;  
GIMMEL'FARB, B.M., redaktor; IVANOV, A.A., redaktor; STRAKHOV, N.M.,  
akademik, otvetstvennyy redaktor; FESENKO, I.A., redaktor; ASTROV,  
A.V., redaktor izdatel'stva; AUZAN, N.P., tekhnicheskiy redaktor

[Problems in the geology of agronomic minerals] Voprosy geologii  
agronomicheskikh rud. Moskva, 1956. 239 p. (MIRA 9:11)

1. Akademiya nauk SSSR. Otdeleniye geologo-geograficheskikh nauk  
(Geology, Economic) (Fertilizers and manures)

FESENKO, I.A.; BEYMAN, Ye.G.

Methodological associations help the biology teachers.  
Biol. v shkole no.4:37-38 Jl-Ag '61.

(MIRA 14:7)

1. Novosibirskiy oblastnoy institut usovershenstvovaniya  
uchiteley.

(Novosibirsk Province--Biology--Study and teaching)

BOYKEVICH, Mikhail Ivanovich; KARVATSKIY, S.B., inzh., retsenzent; YESENKO,  
I.A., inzh., retsenzent; MARENKOVA, G.I., inzh., red.; KHITROV, P.A.,  
tekhn. red.

[Reception and checking-out of centralized traffic control devices;  
experience of communication workers of the Stalin Railroad] Priemka i  
regulirovka ustroistv dispetcherskoi tsentralizatsii; opyt kollektiva  
sviazistov Stalinskoi dorogi. Moskva, Vses. izdatel'sko poligr. ob"edi-  
enie M-va putei soobshcheniya, 1961. 29 p. (MIRA 14:7)  
(Railroads—Signaling—Centralized traffic control)  
(Railroads—Electronic equipment)

"APPROVED FOR RELEASE: 08/23/2000

CIA-RDP86-00513R000412920006-7

FESENKO, I.F., ofitser

Transverse azimuthal scale. Vest. Vozd. Fl. 37 no.1:87 J '55.  
(MIRA 16:8)

APPROVED FOR RELEASE: 08/23/2000

CIA-RDP86-00513R000412920006-7"

FESHENKO, I.F.

Experience in carrying out antituberculosis measures in  
Namangan Province, Uzbek S.S.R. Probl. tub. 34 no.1:7-8 Ja-F '56

1. Iz Oblastnogo protivotuberkuleznogo dispensera v Namangane.  
(TUBERCULOSIS, prev. and control  
in Russia, Uzbek SSR) (MIRA 9:5)

FESENKO, I.F.

Work of a collective farm tuberculosis sanatorium. Sbor. trud. Uz.  
nauch.-issak. tub. inst. 3:196-198 '57. (MIRA 14:5)  
(TYURIA-KURGAN DISTRICT--TUBERCULOSIS--HOSPITALS AND SANATORIUMS)

FESENKO, I.F.

Tuberculosis care in Namangan Province. Med. zhur. Uzb. no.6:  
15-16 Je '58. (MIRA 13:6)

1. Zaveduyushchiy oblastnym tuberkuleznym dispanserom Namanganskoy oblasti.

(NAMANGAN PROVINCE--TUBERCULOSIS)

ALIMOV, Sh. A., prof.; VOLOKHVIANSKIY, A.M., kand. med. nauk; FESENKO, I.F.

Collective farm tuberculosis sanatoria in the Uzbek S. S. R. Probl.  
tub. 36 no.8:3-5 '58. (MIRA 12:7)

1. Iz Uzbekskogo nauchno-issledovatel'skogo tuberkuleznogo instituta  
(dir. Sh. A. Alimov).  
(UZBEKISTAN--TUBERCULOSIS--HOSPITALS AND SANATORIUMS)

FESENKO, I.I.

Effect of the surgical treatment of thyrotoxicosis on the basal metabolism and electrical sensitivity of the eye [with summary in English]. Vrach.delo no.9:105-107 8 '62. (MIRA 15:8)

1. Kafedra obshchey khirurgii (zav. - dotsent K.A.Muzyka) Luganskogo meditsinskogo instituta.  
(HYPERTHYROIDISM) (BASAL METABOLISM) (CHRONAXIA) (EYE)

BELOUS, I.P. [Bilous, I.P.], red.; BOGDANOV, O.P. [Bohdanov, O.P.], red.;  
GUCHEK, I.V. [Huchek, I.V.], red.; MARCHENKO, I.K., red.; SIROTA,  
H.I., red.; STEPANOV, T.K., red.; FEDCHUN, O.K., red.; FESENKO,  
I.K., red.; SLUCHANSKIY, Sh. [Sluchans'kyi, Sh.], tekred.

[The economy of Chernovtay Province; statistical collection]  
Narodne hospodarstvo Chernivets'koi oblasti; statystichnyi  
zbirnyk. Chernivtsi, 1959. 171 p. (MIRA 13:6)

1. Chernovtay (Province) Oblastnoye statisticheskoye upravleniye.  
(Chernovtay Province--Economic conditions)

FESENKO, I.P.

Use of linetol for treating cerebral atherosclerosis under  
polyclinical conditions. Sov. Med. 27 no.7:93-94 Jl '63.  
(MIRA 16:9)

1. Iz poliklinicheskogo otdeleniya Chernigovskoy oblastnoy  
bol'nitsy (glavnyy vrach N.M.Kononenko)  
(CEREBRAL ARTERIOSCLEROSIS) (LINSEED OIL—THERAPEUTIC USE)

FESENKO, I.S., polkovnik meditsinskoy sluzhby

Case of gigantic urterocele. Zdrav.Bel. 8 no.5:55-57 My '62.  
(MIRA 15:10)

FESENKO, L.M.

Changes in the basal metabolism of patients with breast cancer.  
Vop. onk. 8 no.12:78-80 '62. (MIRA 17:6)

1. Iz klinicheskogo otdeleniya Rostovskogo gosudarstvennogo nauchno-issledovatel'skogo instituta rentgenologii, radiologii i onkologii Ministerstva zdravoutdeleniya RSFSR (dir. - P.N. Snegirev.)

9.2540

S/196/61/000/012/019/029  
E194/E155

AUTHOR: Fesenko, M.N.

TITLE: The use of semiconductor devices for voltage control  
of d.c. and a.c. generators

PERIODICAL: Referativnyy zhurnal, Elektrotehnika i energetika,  
no. 12, 1961, 13, abstract 12I 91. (Vestn.  
elektroprom-sti, no. 7, 1961, 40-44)

TEXT: Semiconductor self-excitation systems for d.c. and  
a.c. generators are considered, also the design of the components.  
A method proposed for designing a transistor field circuit can be  
used to construct the generator characteristics and to determine  
the control signal of the transistor which ensures self-  
excitation and that the voltage is maintained constant. It  
follows from comparison of characteristics and from the  
calculations that the inclusion of a transistor introduces no  
major change into the process of self-excitation of a generator.  
A circuit is given of a measuring device in the form of non-  
linear bridges using transistors and a silicon stabililtron;

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1/C

The use of semiconductor devices...

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E194/E155

this differs from existing circuits in that it contains not one but two non-linear elements. The use of a silicon stabililtron in the measuring device reduces to a minimum the effects on the characteristics of the measuring device of changes in transistor properties that result from changes in ambient temperature. Moreover, as the change in control signal is step-wise, there is a great increase in sensitivity of the measuring device. A procedure proposed for designing the measuring device ensures good agreement with experiment. Several possible schematic diagrams for contactless voltage regulators for d.c. and a.c. generators are considered. The voltage regulators which are developed for a.c. differ from existing ones in that the control device is a three-phase controlled rectifier using semiconductor diodes and transistors, which is simultaneously controlled by an electromagnetic relay or non-linear bridge with controlled transistors. Operation of the contactless voltage regulator was checked on a model of a voltage regulator for a d.c. generator type ПЧ-28.5 (PN-28.5) of 3 kW, 110 V. It was found that when rated load was switched on or off the transient process was

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E194/E155

completed in 0.04 - 0.06 seconds. Voltage pulsation on the generator terminals, without the use of filters, did not exceed  $\pm 5\%$  of rated voltage; the accuracy with which the voltage was maintained constant was within  $\pm 0.5 - 1\%$ . Change in ambient temperature over a wide range ( $\pm 50^{\circ}\text{C}$ ) had no important influence on the accuracy of voltage control or the character of the processes.

8 figures. 2 literature references.

[Abstractor's note: Complete translation.]

VC

Card 3/3

*FESENKO, M.N.*

SOV/110-58-9-2/20

AUTHORS: Zdrok, A.G. (Candidate of Technical Science) and  
Fesenko, M.N. (Engineer)

TITLE: The use of Transistors in Voltage-regulator circuits  
(O primenenii kristallicheskikh triodov v skhemakh  
regulirovaniya napryazheniya)

PERIODICAL: Vestnik Elektro promyshlennosti, 1958, Nr 9, pp 4-9 (USSR)

ABSTRACT: Now that high-power germanium rectifiers are being manufactured, voltage regulators based on semi-conducting amplifiers can be made. Current/voltage curves for triode types P4 and P8 are given in Fig 1. The internal resistance of high-output germanium triodes varies over wide limits and they can be used to control the excitation of a generator by acting as variable resistors. Voltage-control circuits for d.c. generators with independent triode control are shown in Fig 2; the transistors are used as variable resistors in the field circuit. An expression is derived for the maximum power output of the transistors. These circuits are suitable for manual control; other circuits are required for automatic control. Although vibration-type regulators have a number of defects they

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SOV/110-58-9-2/20

· The use of Transistors in Voltage-regulator Circuits

are quick and accurate in operation. By using a transistor as the controlling element, the good qualities of vibrating regulators can be taken advantage of by using them as measuring devices. An automatic voltage-control circuit using transistors actuated by a vibrating regulator is shown in Fig 3; here a polarising relay is used as the vibrating regulator. However, it is shown that the transistors currently available can be used for direct control of field current only in low-voltage generators. There are a number of contactless automatic voltage-control circuits. In those shown in Figs 2 and 3 the potential difference between the emitter and the base must alter with the armature speed or the load. This can be effected by a measuring device embodying a transistor and three linear resistors, connected as shown in Fig 4. A graph of the relationship between control and output voltage for this circuit with given values of the different resistors is given in Fig 5. If this measuring device is connected to the generator output terminals and the output is applied to a rectifier in the field circuit, the generator voltage can be made

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**The use of Transistors in Voltage-regulator Circuits**

stable when the armature speed or load alters. A circuit of this kind is seen in Fig 6 and its operation is explained. As the voltage on the generator terminals increases, the output voltage of the measuring device is reduced; this increases the internal resistance of the corresponding triode, so reducing the field current. The applicability of circuits embodying transistors may be extended by the use of amplidyne. Fig 7 shows an automatic voltage-regulation circuit for a d.c. generator. The field winding is supplied with rectified a.c. An amplidyne is connected in the a.c. input circuit and the measuring circuit shown in Fig 4 is applied to the d.c. winding. An automatic voltage-control circuit suitable for an a.c. generator is shown in Fig 8; it was used to control the output of a three-phase 10 kVA, 230 V generator. It is concluded that transistors can be widely used in automatic voltage-control circuits for d.c. and a.c. generators. Automatic voltage-controllers based on transistors will be lighter, smaller, faster and more reliable than those now in use. The best results

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SOV/110-58-9-2/20

• The use of Transistors in Voltage-regulator Circuits

are obtained by combining transistors with magnetic and electro-magnetic amplifiers. An appendix includes test results on a voltage-control circuit including a vibrating regulator (Fig 3) for a d.c. generator of 1.5 kW 28 V and of a contactless regulator (Fig 6) for an automobile generator of 350 W and 14 V. Oscillograms illustrating the operation of the voltage-regulator with vibrating control are reproduced in Fig 9, for various conditions of operation. Oscillograms of the contactless regulator appear in Fig 10. These oscillograms and the results of a number of other investigations show that when

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SOV/110-58-9-2/20

The use of Transistors in Voltage-regulator Circuits

the load is raised from zero to the rated value the output voltage falls by about 2%. As the generator speed rises from 1200 to 3500 r.p.m. the output voltage alters by 3 - 5%.

There are 10 figures, and 6 references, 4 of which are Soviet and 2 English.

SUBMITTED: March 24, 1958

1. Voltage regulator--Circuits
2. Transistors--Applications
3. Voltage regulators--Performance
4. Control systems--Applications

Card 5/5

FESENKO, M. N., kand.tekhn.nauk (Moskva)

Method for cutting off the forcing of a transient.  
Elektrichestvo no.6:58-61 Je '60. (MIRA 13:7)  
(Electric driving)  
(Rolling mills--Electric driving)

L 36286-65 EWT(1)/EEC(1)-2/EWG(1)/T/EEC(1)-2/EW(1) Pm-4/Pz-6/Pet IJP(c)

ACCESSION NR: AP5008164

S/0286/65/000/005/0039/0040

AUTHOR: Fesenko, M. N.

TITLE: A controlled rectifier using a semiconductor triode. Class 21, No. 168759

SOURCE: Byulleten' izobreteny i tovarkh zhakov, no. 5, 1965, 39-40

TOPIC TAGS: rectifier, signal shaping, phase inverter

ABSTRACT: This Author Certificate presents a controlled rectifier using a semiconductor triode with a control signal shaping unit and a regulated phase inverter included in the triode control circuit (see Fig. 1 on the Enclosure). The design is intended to increase the rectifier efficiency and to simplify the control signal shaping unit. A ladder network, consisting of a limiting resistor and a bilateral clipper made in the form of anti-parallelly connected diodes, is connected between the phase inverter and the input of the semiconductor triode, serving as the shaping unit. Orig. art. has 1 figure.

ASSOCIATION: none

SUBMITTED: 27Nov61

ENCL: 01

SUB CODE: EC

NO REF SOV: 000

OTHER: 000

Card 1/2

L 177990-66

ACI NRI AP6006338

SOURCE CODE: UR/0413/66/000/002/0059/0059

32  
B

INVENTOR: Fesenko, M. N.

ORG: none

TITLE: Reversible composite transistor. Class 21, No. 177990

SOURCE: Izobreteniya, promyshlennyye obraztsy, tovarnyye znaki, no. 2, 1966, 59

TOPIC TAGS: transistor, transistorized circuit

ABSTRACT: The reversible composite transistor shown in Fig. 1 consists of two PNP power transistors connected in parallel but with the collector of one connected to

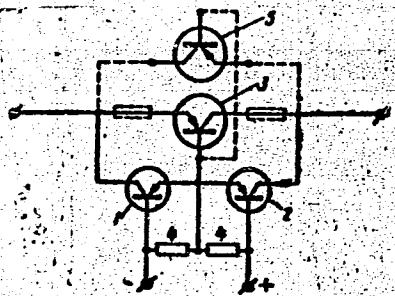


Fig. 1 Composite reversible transistor

1, 2 - Control transistors;  
3, 5 - power transistors;  
4 - resistors.

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UDC: 621.382.333.34

2

L 17690-66

ACC NR: AP6006338

the emitter of the other. To control the operating mode of the composite power transistor, two control transistors are added, one shunting the collector-base and the other, the emitter-base junction of the power transistor. Both emitters of the control transistors are connected to the bases of the power transistors. Two resistors or diodes are placed between the bases of the control transistors and the bases of the power transistors. The compound reversible transistor obtained in this manner is symmetrical. Orig. art. has: 1 figure. [BD]

SUB CODE: 09/ SUBM DATE: 16Jul64/ ATD PRESS: 4209

Card 2/2

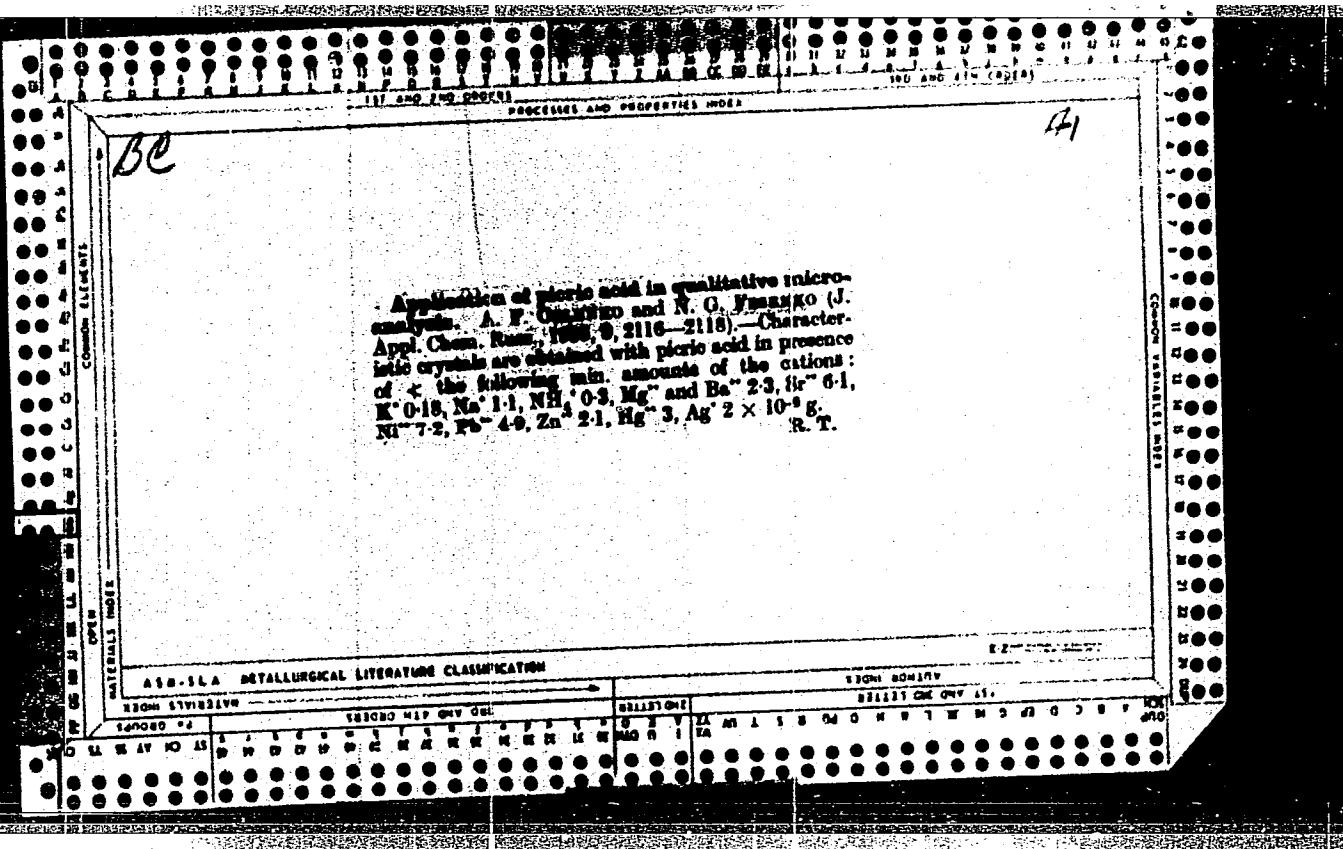
DATSKO, V.G., doktor khim. nauk; PONOMAREV, N.F., doktor khim. nauk; FESENKO,  
N.G., kand.khim. nauk; BRAZHNICKOVA, L.V., kand.khim.nauk

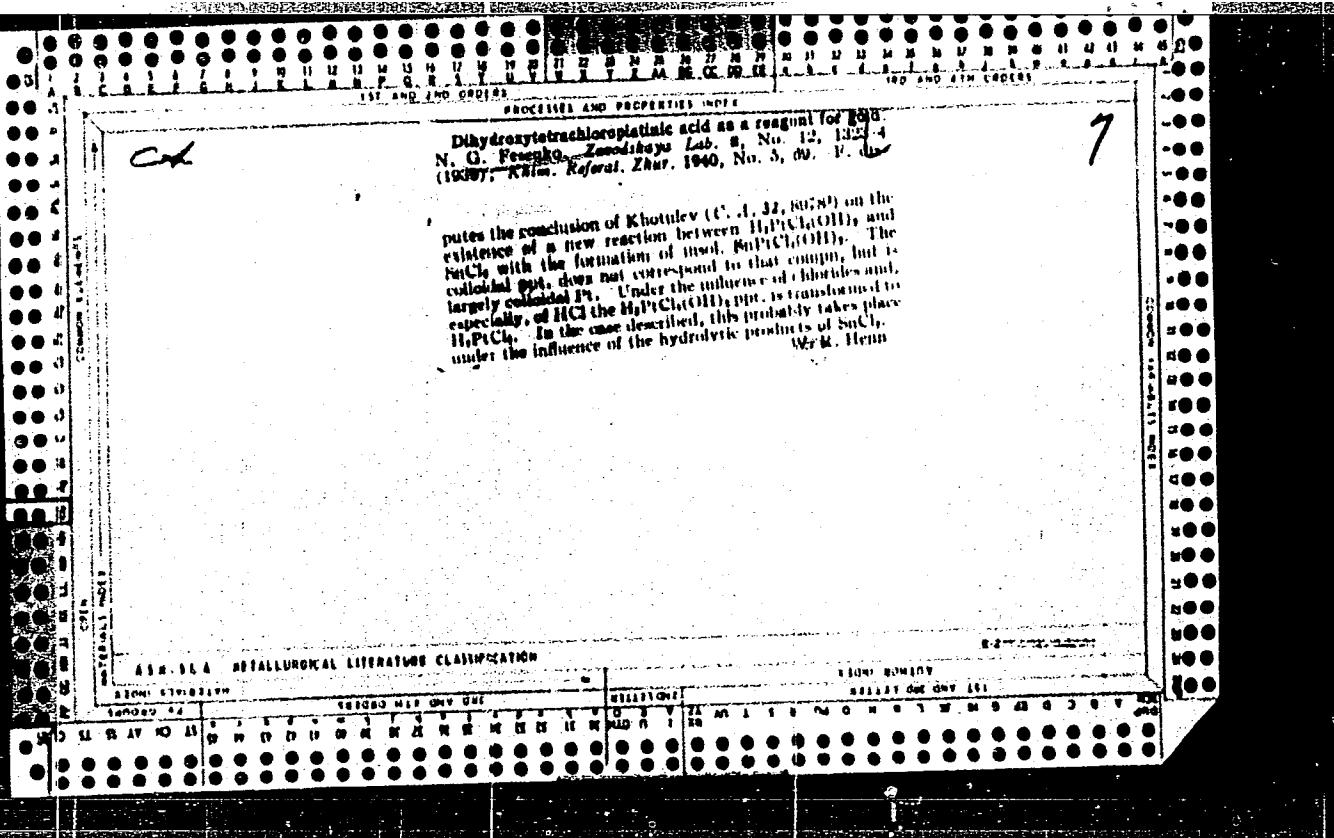
The 17th hydrochemical conference. Zhur. VKHO 8 no.6:695 '63.

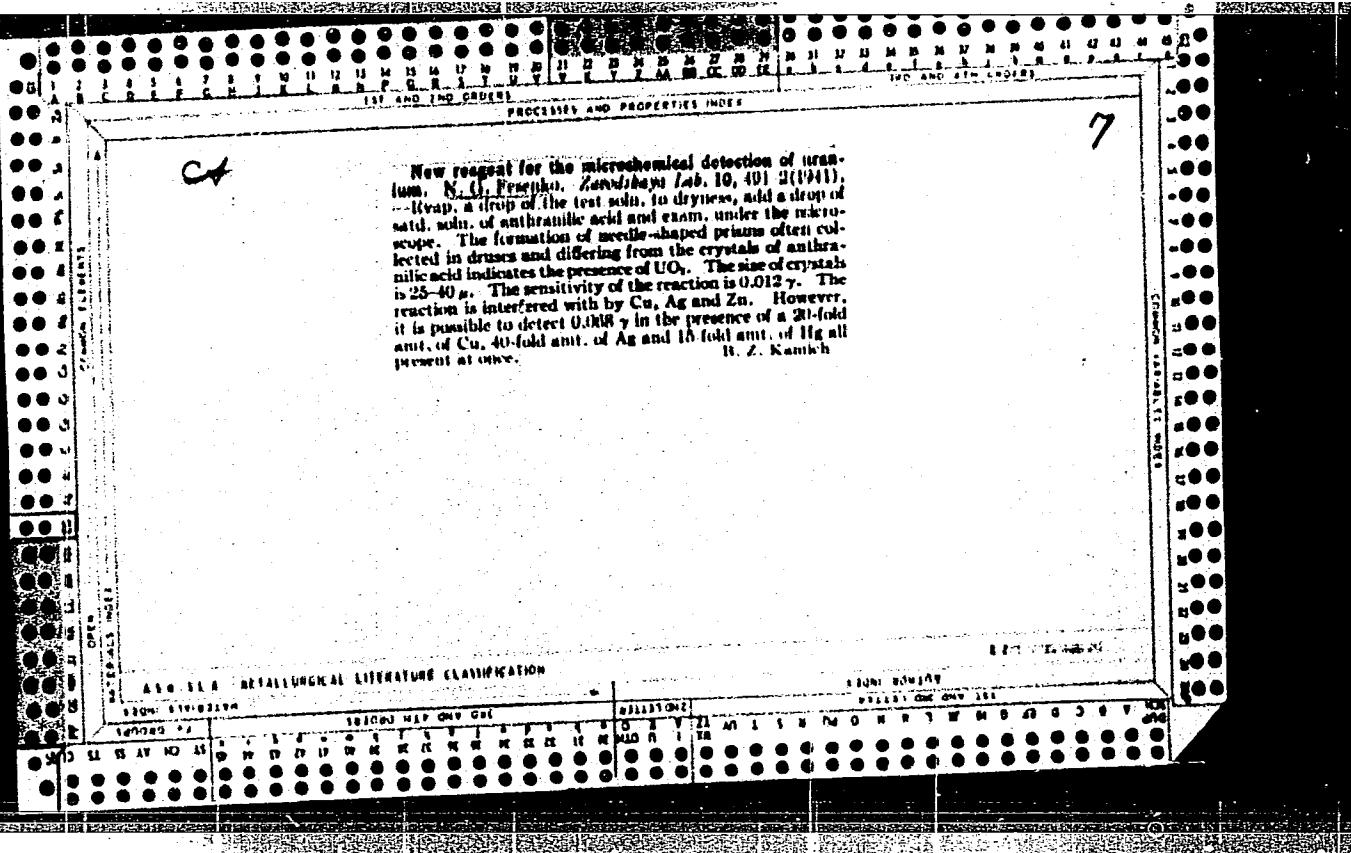
(MIRA 17:2)

SOLOMIN, Gennadiy Anatol'yevich; FESENKO, N.G., kand. khim. nauk,  
otv. red.; DRAGUNOV, E.S., red.

[Methods for determining the redox potential and pH of  
sedimentary rocks] K metodike opredelenija okislitel'no-  
vosstanovitel'nogo potentsiala i pH osadochnykh porod.  
Moskva, Izd-vo "Nauka," 1964. 86 p. (MIRA 17:7)







CA

Mixer for substances with high viscosity. N. G. Kozak.  
cake. (Acad. Sci., U.S.S.R.). Zvezdochka Lab. 13,  
13(1949).—The mixer is an electrically driven perforated  
plate traveling in a vertical cylinder containg the  
sample. A packing gland for the connecting rod insures  
protection of sample from loss or contact with the atm.  
G. M. Kondopod

*C.A.**14*

Foaming of boiling mixtures of electrolytes which predominate in boiler water. N. G. Fesenko. *Doklady Akad. Nauk S.S.R.*, 72, 547-50 (1950)—Intensity of foaming of  $\text{NaCl-Na}_2\text{CO}_3\text{-NaOH}$  and  $\text{Na}_2\text{SO}_4\text{-Na}_2\text{CO}_3\text{-NaOH}$  was measured by the amt. of ml. of liquid carried over with the foam during rapid decrease of pressure over the boiling soln. Spatial diagrams were constructed showing foaming as a function of compn. and total concn. of each system. These make it possible to det. the foam carry-over of any mixt. of electrolytes with a total concn. of 20-120 milliequiv./liter. This cannot be directly applied for evaluating the foaming of water of the same compn. in a b'ler because of the design features and operating conditions of the b'ler. However, after the foaming intensity in a com. installation is detd. for several concns. corresponding to different fields of these diagrams, it is possible to forecast the foaming of the waters with changes in compn.—of dissolved electrolytes. This possibility of forecasting foaming is of great importance for stationary boilers and for railway transport. The diagrams also show that the foaming action of mixts. of electrolytes does not follow the principle of additivity. The time for blowing the boilers to reduce foaming should be detd. not on the basis of  $\text{Cl}^-$  or total salt content but on the basis of the content of  $\text{Cl}^-$ ,  $\text{SO}_4^{2-}$ ,  $\text{CO}_3^{2-}$ , and  $\text{OH}^-$  in the waters. B. Z. K.

FESENKO, N. G.

PA 193T29

USSR/Chemistry - Foam-Formation in Boilers

Oct 51

"Evaluation of Blowing of Boilers as a Method for Combating Foaming of Boiler Water," S.A. Durov, Ya. M. Nemirovskiy, N.G. Fesenko, Hydrochem Inst, Acad Sci USSR

"Zhur Prik Khim" Vol XXI, No 9, pp 989-992

Investigation of ability of boiling solns to form foam by method of foam entrainment shows that mixts of electrolytes act much more strongly than calcns by rule of additivity show. Inorg colloids with both pos and neg charges have foam-forming action. Constructed diagram of form entrainment for ternary system of electrolytes characteristic for boiler water in iron boilers by chloride or total salt content in boiler water must be replaced by more rational detn of tendency toward foam-formation from diagrams of ternary (or quaternary) systems.

PA 193T29

P. TESLINSKO, N. 6

The problem of mineral content of water in the Volga River seems to be one of the most important problems in the development of the Volga basin.

Water from the Volga River contains a large amount of minerals.

Mineral content of water in the Volga

River is determined by the following factors:

1. Mineral content of the soil.

2. Mineral content of the rocks.

3. Mineral content of the air.

4. Mineral content of the water in the

Volga River.

5. Mineral content of the water in the

Volga River.

6. Mineral content of the water in the

Volga River.

7. Mineral content of the water in the

Volga River.

8. Mineral content of the water in the

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9. Mineral content of the water in the

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10. Mineral content of the water in the

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33. Mineral content of the water in the

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34. Mineral content of the water in the

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35. Mineral content of the water in the

Volga River.

36. Mineral content of the water in the

Volga River.

FESENKO, N.G.

Foaming of boiling solutions of electrolytes NaCl - Na<sub>2</sub>SO<sub>4</sub> -  
(mixture of 50% NaOH and 50% Na<sub>2</sub>CO<sub>3</sub>) and NaCl - Na<sub>2</sub>SO<sub>4</sub> -  
NaOH. Gidrokhim.mat. no.20:120-127 '53. (MIRA 7:3)

1. Gidrokhimicheskiy institut Akademii nauk SSSR, Novocherkassk.  
(Foam) (Electrolytes)

EFENKO, N. G.

the complexometric determination of calcium in natural waters with the aid of murexide as indicator

Fisenko (Hydrochim. i. zerklyaniia)

January 21, 1957 (1955) 11(1) 137-142  
Water containing less than 40-50 mg. Ca/l in an Erlenmeyer flask, add dry mixed indicator (urethane 1.0 and NaCl 0.5 g.) and 2 cc. of 2N NaOH. After a short time add 1 cc. of dilute nitrate with 0.01N Toluol B-1 until a change in color of violet which persists for 1 min.  $\lambda = 540 \text{ m} \mu$  (at  $t = 25^\circ \text{C}$  and 1000, V), where  $t$  is normality of I,  $\lambda$  - c. of I, and  $V$  the vol taken for titration. Cu, Ag, and Zn interfere with the reaction. To remove Cu add 1-2 drops of 5% NaOH and 1 drop of ammonia, that of Mn by a few drops of 5% NH<sub>4</sub>OAc/ICl. Cu, Zn and Fe are removed with a few drops of Na<sub>2</sub>S. The synthesis of murexide from uric acid through the dioxan and alkoxanth stages is described.

A. S. Mackin

14-57-6-12301

Translation from: Referativnyy zhurnal, Geografiya, 1957, Nr 6,  
p 84 (USSR)

AUTHORS: Fasenko, N. G., Denin, A. A.

TITLE: Chemical Composition of Water in the Main Don Canal  
and the Lower Don Canal (O khimicheskem sostave vody  
Donskogo magistral'nogo i Nizhne-Donskogo kanalov)

PERIODICAL: Gidrokhim. materialy, 1955, Vol 25, pp 170-175

ABSTRACT: Investigations carried out by the authors in 1952 and  
1953 have shown that water in the canals during the  
first months of their use differed from the water in  
the Teimlyanskoje reservoir which supplies them by  
higher mineral content; this is caused by leeching  
of easily soluble salts from the canal bed. The  
content of principal ions has not changed since  
August 1952 through the length of the canal, and during  
the year the ion content has changed in the same  
way as the content in the water at the lower part of

Card 1/2

14-57-6-12301

Chemical Composition of Water (Cont.)

the Tsimlyanskoye reservoir. The canal water belongs to the bicarbonate calcium group of type II; it is characterized by the absence of  $\text{CO}_3^{2-}$ , and by the small amount of ions  $\text{Na}^+$ ,  $\text{Cl}^-$  and  $\text{SO}_4^{2-}$ .

$\text{SO}_4^{2-}$ .

Card 2/2

O. V. B.

FESENKO, N.G.

Hydrochemical character of the Tsimlyansk Reservoir during its  
initial period of operation. Gidrokhim.mat.25:69-97 '55.

(MIRA 9:6)

1.Gidrokhimicheskiy institut Akademii nauk SSSR, Novocherkassk.  
(Tsimlyansk Reservoir--Water)

FESENKO, N.G.

USSR/Cosmochemistry - Geochemistry. Hydrochemistry, D

Abst Journal: Referat Zhur - Khimiya, No 19, 1956, 61353

Author: Fesenko, N. G., Rogozhin, V. I., Fesenko, Ye. A., Sheynin, M. S.

Institution: None

Title: Prevalent Conditions of Dissolved Gases and Hydrobiology of the Tsimlyanskoye Reservoir during the Early of the First Winter Stagnation

Original  
Periodical: Gidrokhim. materialy, 1955, 25, 98-114

Abstract: The first 1952-1953 winter period in the history of Tsimlyanskoye reservoir was characterized by a sufficiently high content of dissolved oxygen in the water from beginning to the end of the ice-bound period. This high O<sub>2</sub> content was due during the initial period the intensive wind-induced aeration of the water and persisted thereafter as a result of low temperatures of the water in conjunction with paucity ~~of~~ zooplankton and benthos. Small depth of the snowcover could contribute to production of O<sub>2</sub> as a result

Card 1/2

USSR/Cosmochemistry - Geochemistry. Hydrochemistry, D

Abst Journal: Referat Zhur - Khimiya, No 19, 1956, 61353

Abstract: of life activities of phytoplankton, but with a small amount of biomass of the latter the quantity of phylogenetic O<sub>2</sub> could not be considerable and was probably depleted by O<sub>2</sub> consumption of the zooplankton. Retention of a relatively high O<sub>2</sub> content was also sustained by a rise of the water level in the reservoir during the icebound period which prevents the discharge into the reservoir of ground waters poor in oxygen. Dynamics of vertical distribution of O<sub>2</sub> is dependent upon the nature of the submerged vegetation.

Card 2/2

FESENKO, N.G.; ZHIN, A.A.

Chemical composition of water in the main Don Canal and in the lower Don Canal. Gidrokhim. mat. 25:170-175 '55. (MIRA 9:6)

1.Gidrokhimicheskiy institut Akademii nauk SSSR, Novocherkassk.  
(Volga-Don Canal--Water)

FESENKO, N.G.

USSR/Cosmochemistry - Geochemistry. Hydrochemistry, D

Abst Journal: Referat Zhur - Khimiya, No 19, 1956, 613-8

Author: Fesenko, N. G., Zenin, A. A.

Institution: None

Title: On Chemical Composition of Azov Main Canal and Questions Relating to Its Formation

Original  
Periodical: Gidrokhim. materialy, 1955, 25, 178-182

Abstract: Field investigations during 1952-1953 revealed that content of principal ions in water of canal is distributed unevenly along its length but no over-all increase in total ions was noted. Mineralization of water during vegetative period varies but slightly (1,327-1,408 mg/l in 1952, 1,259-1,479 mg/l in 1953, for water inflowing to the canal). In all samples of water is found an excess of Mg over Ca. According to classification of Aleksin the canal water appertain sometimes to sulfate, sometimes to chloride class, and sodium group of second type. Irrigational coefficient

Card 1/2

*Geochem. Inst. Novocherkassk*

USSR/Cosmochemistry - Geochemistry. Hydrochemistry, D

Abst Journal: Referat Zhur ~ Khimiya, No 19, 1956, 613-8

Abstract: of the water is 5.7-7.4 which constitutes a hazard of soil salination. Sources of accumulation of principal ions are under [redacted] waters of zone of active water exchange between river B. Yegorlyk and Veselovskiy reservoir. Nevinnomysskiy canal exercises little influence on composition of water in the main canal.

Card 2/2

FESENKO, N.G.

Direct trilonometric determination of the di- and trivalent iron dissolved in water. Gidrokhim. mat. 27:135-139 '57. (MIRA 11:4)

1. Gidrokhimicheskiy institut AN SSSR, Novocherkassk.  
(Iron--Analysis) (Titration)

FESENKO, V.G.

Direct trilonometric determination of magnesium in water. Gidrokhim.  
mat. 27:140-145 '57. (MIRA 11:4)

1. Gidrokhimicheskiy institut AN SSSR, Novocherkassk.  
(Magnesium--Analysis) (Titration)

FESENKO, N. G.

50-2-14/22

AUTHOR: Fesenko, N. G.

TITLE: Water Sampling Device for River Cross Sections (Prisposobleniye dlya otbora prob vody po poperechnomu stvoru reki)

PERIODICAL: Meteorologiya i Gidrologiya, 1958, Nr 2, pp. 44 - 44 (USSR)

ABSTRACT: Differences in the percentage of solved chemicals will occur in various points of the river usually after the junction with its tributary, the inflow of industrial waste water, and others. The influence of the inflows is so great that one was forced to take a series of samples in several depths of the water in order to make up a characteristic. For the water sampling from the North-Donez a device was used which can be used successfully also for investigations of other not navigable rivers. This device has proved to be very comfortable and permits an automatic sampling in the desired depth (figure 1). A rubber hose is fastened by means of a metal ring to a wire attached to blocks; the end of the rubber

Card 1/2

50-2-14/22

Water Sampling Device for River Cross Sections

hose is fastened to a shaft hanger fitted out with a weight. A displacement of the wire causes the displacement of the weight and of the end of the rubber hose. The existence of marks on the wire permits to put the end in precisely the desired vertical position. The sampling is carried out by means of a pump, or according to the principle of the siphon, undisturbedly a more simple way. The described device permits to take quickly and in precisely the given horizons in the course of 24 hours of observation, also at night or during a thunder-shower. There is 1 figure.

AVAILABLE: Library of Congress

Card 2/2

AUTHOR:

Fesenko, N.G.

32-1-9/55

TITLE:

The "Trilonometric" Determination of Iron in Ores and Agglomerates (O trilonometricheskem opredelenii zheleza v rudakh i agglomerate).

PERIODICAL:

Zavodskaya Laboratoriya, 1958, Vol. 24, Nr 1, pp. 25-26 (USSR)

ABSTRACT:

Quantitative determinations of this kind were suggested by Usatenko and Mikhaylova [Ref. 1], viz. by the titration of iron with a trilon-B-solution in sulfosalicylic acid as indicator. A renewed examination of this method gave results which were higher by 5%, which is in this case explained by the fact that "cold titration" was used. Titration at 60° was recommended by Flaschka [Ref. 2], Lyndersen and Gjems [Ref. 3], as well as by Bashkirtseva and Yakimets [Ref. 4]. In the case of the method developed by Usatenko and Mikhaylova it is further criticized that it does not take pH-value control into account, whereas, as is maintained, reliable results would be obtainable only at pH-values=1-3. According to Kuznetsov [Ref. 5] disturbing colorings may form with other ions with pH=4-4.5. A process of analysis is described in the paper, by means of which these shortcomings could be removed.

Card 1/2

The "Trilonometric" Determination of Iron in Ores and  
Agglomerates

32-1-9/55

The sample is here dissolved in a hydrochloric acid solution and is diluted with water. Into the solution a strip of Congo indicator paper or a tropeolin strip (OO) is introduced; furthermore, nitric acid is added in drops to the solution until the red color of the Congo paper becomes bluish-violet ( $\text{pH}=3$ ), or, in the case of tropeolin, it turns from yellow to red ( $\text{pH}=1,3$ ). The solution is then heated up to  $60^{\circ}$ ; for this purpose some crystals of sulfo-salicylic acid are added and titration with a trilon B solution is carried out until the violet coloring disappears. There are 6 references, 4 of which are Slavic.

ASSOCIATION: Hydrochemical Institute AN USSR (Gidrokhimicheskiy institut Akademii nauk SSSR).

AVAILABLE: Library of Congress

Card 2/2      1. Iron-Determination    2. Titration

FESENKO, N.G.

Extraction method for purifying ground material is one of the ways to  
reduce sewage contamination. Gig. i san. 24 no.9:78-79 S '59.

(MIRA 13:1)

1. Iz Gidrokhimicheskogo instituta Akademii nauk SSSR.  
(EXTRACTION APPARATUS)

KAPLIN, V.T., starshiy laborant; FESENKO, N.G., starshiy nauchnyj sotrudnik,  
kandidat khimicheskikh nauk

Quantitative determination of phenols in natural reservoirs when  
their content is 0.001 mg. per liter and higher. Gig.i san. 25  
no.8:41-43 Ag '60. (MIRA 13:11)

1. Iz Gidrokhimicheskogo instituta AN SSSR.  
(WATER ANALYSIS) (PHENOLS)

PHASE I BOOK EXPLOITATION

SOV/5374

Akademiya nauk SSSR. Gidrokhimicheskiy institut

Gidrokhimicheskiye materialy, t. XXX (Hydrochemical substances, v. 30)  
Moscow, Izd-vo AN SSSR, 1960. 213 p. Errata slip inserted.  
2,000 copies printed.

Sponsoring Agency: Akademiya nauk SSSR. Gidrokhimicheskiy institut  
(Novocherkassk).

Editorial Board (Title page): Resp. Ed.: O. A. Alekin, N. V.  
Veselovskiy, Deputy Resp. Ed. V. G. Datsko, G. S. Konovalov,  
M. I. Kriventsov, P. A. Kryukov, Resp. Secretary and K. G.  
Lazarev. Ed. of Publishing House: D. N. Trifonov. Tech. Ed.:  
I. T. Dorokhina.

PURPOSE: This publication is intended for hydrologists, hydrochemists,  
and hydrometeorologists.

COVERAGE: This is a collection of 22 articles on the hydrochemistry  
of rivers and water bodies in the USSR. The authors discuss

-Card 1/8-

Hydrochemical Substances

SOV/5374

pollution, spectrographic methods of determining the content of microelements in water, and the content and discharge of ions, gases, as well as chemical, biogenic, and organic substances. A map showing the distribution of the ionic discharge of rivers in the USSR is the most complete to appear in print to date. No personalities are mentioned. Each article is accompanied by references.

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Card 2/3

## Hydrochemical Substances

SOV/5374

Krepkogorskiy, L. N. [Kafedra gigiyeny Kazanskogo gosudarstvennogo instituta usovershenstvovaniya vrachey im. Lenina-Department of Hygiene, Kazan' State Institute for the Improvement of Physicians imeni Lenin]. Fluorine in the Surface Waters of Kazakhstan

32

Veselovskiy, N. V., and I. A. Goncharova [Hydrochemical Institute AS USSR]. Regime of Dissolved Gases and Biogenic Substances as Sampled in One of the Ponds of the Rostovskaya Oblast'

43

Fesenko, N. G. [Hydrochemical Institute AS USSR]. Chemical Composition of the Waters of the Severskiy Donets River in Its Area of Greatest Pollution

65

Fesenko, N. G. [Hydrochemical Institute AS USSR]. Phenols in the Waters of the Severskiy Donets and Don Rivers

75

Rozinoyer, I. M. [Kafedra khimii Voronezhskogo Zoovetinstituta - Department of Chemistry, Voronezh Zoological Veterinary

Card 3/8

KAPLIN, V.T.; FESENKO, N.G.

Rapid method of determining ammonium ions in the waste water  
from the manufacture of by-product coking plants. Koks i  
khim. no.5:49-50 '60. (MIRA 13:7)

1. Gidrokhimicheskiy institut AN SSSR.  
(Sewage—Analysis) (Ammonium salts)  
(Coke industry—By-products)

FESENKO, N.G.

Phenols in waters of the Northern Donets and Don Rivers. Gidrokhim.  
mat. 30:75-83 '60. (MIRA 13:9)

1. Gidrokhimicheskiy institut AN SSSR, Novocherkassk.  
(Northern Donets River--Water--Pollution) (Phenols)  
(Don River--Water--Pollution)

FESENKO, N.G.; KLIMOV, I.T.

Amount of heavy metals in the Kázennyy Torets and Northern Donets  
during irrigation periods. Gig. i san. 25 no.3:104-105 Mr '60.  
(MIRA 14:5)

1. Iz Gidrokhimicheskogo instituta Akademii nauk SSSR.  
(DONETS VALLEY-WATER-POLLUTION) (METALS)

ZENIN, A.A.; ROGOZHIN, V.I.; FESENKO, N.G.

Nature of the movement of water masses near the dam in Tsimlyansk,  
Gorkiy, Kuybyshev, and Stalingrad Reservoirs. Gidrokhim. mat. 32:113-  
121 '61. (MIRA 14:6)

1. Gidrokhimicheskiy institut AN SSSR, Novocherkassk.  
(Reservoirs)  
(Hydraulics)  
(Water—Composition)

FESENKO, N.G. (Novocherkassk); SOLOMIN, G.A. (Novocherkassk)

Method for fast voluminal determination of  $\text{Fe}^{++}$ ,  $\text{Fe}^{II}$ , and  $\text{Al}^{+++}$   
in ferric and mixed coagulants. Vod. i san. tekhn. no.1:16-17  
(MIRA 14:9)  
Ja '61.

(Water--Purification)

KLIMOV, I.T., mladshiy nauchnyy sotrudnik; FESENKO, N.G., starshiy nauchnyy sotrudnik, kand.khimicheskikh nauk

Pollution of acid waters by heavy metals in the coal mines of the Donets Basin. Gig. i san. 26 no.5:97-98 My '61. (MIRA 15:4)

1. Iz Gidrokhimicheskogo instituta AN SSSR.  
(DONETS BASIN--WATER--POLLUTION) (MINE WATERS)

KAPLIN, V.T., mladshiy nauchnyy sotrudnik; FESENKO, N.G., starshiy nauchnyy  
sotrudnik, kand.khimicheskikh nauk

Preservation of water samples containing phenols. Gig. i san. 26  
no.6:68-69 Je '61. (MIRA 15:5)

1. Iz Gidrokhimicheskogo instituta AN SSSR.  
(WATER—ANALYSIS) (PHENOLS)

DATSKO, V.G., doktor khim.nauk; FESENKO, N.G., kand.khim.nauk; BRAZHNICKOVA,  
L.V., kand.khim.nauk

Hydrochemical sources for the comprehensive utilization and protection  
of water resources. Vest.AN SSSR 31 no.9:135-136 S '61.  
(MIRA 14:10)

(Water—Analysis)

DATSKO, V.G., prof.; FESENKO, N.G., kand.khimicheskikh nauk; BRAZHNJKOVA,  
L.V.; PONOMAREV, I.F., prof.

Fifteenth All-Union Hydrochemical Conference. Zhur. VKh0 6 no.6:  
702 '61. (MIRA 14:12)  
(Water conservation--Congresses)

KAPLIN, V.T.; FESENKO, N.G.

Determination of phenols in water by means of pyramidon. Zav.lab.  
28 no.3:287-288 '62. (MIRA 15:4)

1. Gidrokhimicheskiy institut AN SSSR.  
(Phenols) (Aminopyrine)

DATSKO, V.G., doktor khimicheskikh nauk; PONOMAREV, I.F., doktor khimicheskikh nauk; FESENKO, N.G., kand.khimicheskikh nauk; BRAZHNICKOVA, L.V., kand.khimicheskikh nauk

Sixteenth Hydrochemical Conference. Zhur. VKHO 7  
no.6:690 '62. (MIRA 15:12)  
(Water--Composition)

DATSKO, V.G., doktor khim.nauk; FESENKO, N.G., kand.khim.nauk; BRAZHNIKOVA,  
L.V., kand.khim.nauk

Studies of the chemical composition of surface waters. Vest.AN  
SSSR 32 no.8:124-125 Ag '62. (MIRA 15:8)  
(Water—Composition)

LEBEDEVA, Ye.M.; FESENKO, N.G.

Hydrochemical regime of the Northern Donets River near the village of Svetlichnoye after the beginning of exploitation of the Northern Donets -- Donets Basin Canal. Gidrokhim. mat. 35:107-115 '63.  
(MIRA 16:7)

1. Gidrokhimicheskiy institut, Novocherkassk.  
(Northern Donets River--Water--Composition)

LEBEDEVA, Ye.M.; FESENKO, N.G.

Pollution map of Donets Basin rivers. Gidrokhim. mat. 35:116-120  
'63. (MIRA 16:7)

1. Gidrokhimicheskiy institut, Novocherkassk.  
(Donets Basin--Water--Pollution)

ZAYATS, G.N.; FESENKO, N.G.

Mine waters of Rostov Province. Gidrokhim. mat. 35:131-134 '63.  
(MIRA 16:7)

1. Gidrokhimicheskiy institut, Novocherkassk.  
(Rostov Province--Mine water--Composition)

BABESHKINA, Z.M.; KAPLIN, V.T.; FESENKO, N.G.

Colorimetrik determination of phenols in water. Gidrokhim. mat.  
35:207-217 '63. (MIRA 16:7)

1. Gidrokhimicheskiy institut, Novocherkassk.  
(Water--Composition) (Phenols)

FESENKO, N.C., ZENIN, A.A.

Change in the mineralization of groundwater in the backwater  
area of the Tsimlyansk Reservoir. Gidrokhim. mat. 37:56-62  
'64. (MIRA 18:4)

1. Gidrokhimicheskiy institut Glavnogo upravleniya gidro-  
meteorologicheskoy sluzhby pri Sovete Ministrov SSSR, Novo-  
charkassk.

FESENKO, N.G.

Problems of hydrochemistry in the field of conservation of  
natural waters. Gidrokhim. mat. 37:125-132 '64. (MIRA 18:4)

1. Gidrokhimicheskly institut Glavnogo upravleniya gidrometeoro-  
logicheskoy sluzhby pri Sovete Ministrov SSSR, Novocherkassk.

KAPLIN, V.T.; SOLGIN, G.A.; FESENKO, N.G.

Character of the water pollution of the Volgograd Reservoir  
within Saratov Province under flood conditions. Gidrokhim.  
mat. 37:144-147 '64. (MIRA 18:4)

1. Gidrokhimicheskiy institut Glavnogo upravleniya gidrometeorologicheskoy sluzhby pri Sovete Ministrov SSSR, Novecherkassk.

BEYSOVA, M.P.; SOLOMIN, G.A.; FESENKO, N.G.

Determining the acidity of mine waters. Gidrokhim. mat. 37:  
148-153. '64. (MIRA 18:4)

1. Gidrokhimicheskiy institut Glavnogo upravleniya gidrometeoro-  
logicheskoy sluzhby pri Sovete Ministrov SSSR, Novocherkassk.

ZAVODNOV, S.S.; SOLOMIN, G.A.; FESENKO, N.G.

Neutralization of acid waste water in intermediate ponds.  
Gidrokhim. mat. 37:154-157 '64. (NIRA 18:4)

1. Gidrokhimicheskiy institut Glavnogo upravleniya gidrometeorologicheskoy sluzhby pri Sovete Ministrov SSSR, Novocherkassk.

KAPLIN, V.T.; FESRENKO, N.G.; BABESHKINA, Z.M.; SIMIRENKO, V.I.

Effect of temperature on the disintegration rate of monatomic phenols in natural waters. Gidrokhim. mat. 37:152-163 '64.  
(MIRA 18:4)

1. Gidrokhimicheskiy institut Glavnogo upravleniya gidrometeorologicheskoy slushby pri Sovete Ministrov SSSR, Novecherkassk.

FESENKO, N.G.

Some ways of regulating the quality of water by means of  
the supplementary elements of water-management apparatus.  
Gidrokhim.mat. 36:182-186 '64.

(MIRA 18:11)

1. Gidrokhimicheskiy institut, Novocherkassk. Submitted  
August 30, 1961.

LEBEDEVA, Ye.M.; FESENKO, N.Q.

Effect of the "Donsoda" combine on the mineralization of water  
in the Northern Donets River. Gidrokhim.mat. 36:64-74 '64.  
(MIRA 18:11)

1. Gidrokhimicheskiy institut, Novocherkassk. Submitted  
December 16, 1961.

ZAVUDNOV, S.S.; FESENKO, N.G.

Value of the first constant of Mn ion hydrolysis and the  
 $Mn(OH)_2$  solubility product. Gidrokhim.mut. 36:148-155 '64.  
(MIRA 18:11)

1. Gidrokhimicheskiy institut, Novocherkassk. Submitted  
December 15, 1961.

POPOV, R.I.; FESENKO, N.I.; SIPOVICH, S.Yu.; SHELKOV, S.K.

Continuous fusion of sulfur. Koks. i khim. no. 3:46-48 '61.  
(MIRA 14:4)

1. Dnepropetrovskiy koksokhimicheskiy zavod.  
(Dnepropetrovsk—Coke industry—By-products)

FESENKO, N.N., kapitan 3-go ranga

Errors of distant reading gyroscopic compasses of the "Kurs"  
and "Oiria" type. Mer. sbor. 48 no.10:56-61 O '65. (MIRA 18:9)

(N) 1 11928-66 EWT(d) BC  
ACC NR: AP6001834 SOURCE CODE: UR/0375/65/000/010/0056/0061

44,55  
AUTHOR: Fesenko, N. N. (Lieutenant commander)

ORG: none

TITLE: Error of the repeaters of the "Kurs" and "Girya" gyrocompasses

SOURCE: Morskoy sbornik, no. 10, 1965, 56-61

TOPIC TAGS: ship navigation, error prediction, error correction, gyrocompass

ABSTRACT: After noticing on one of the ships that the gyrocompass correction is not constant in magnitude and in sign, the author established, by analyzing the errors, that the gyrocompass correction depends on the course angle and the magnitude of the compass direction; i.e., it is burdened by the total error of two components and is characteristic of this type of repeater. The error appears usually after the change of glass or of the compass card for the coarse reading of the repeater and insufficient centering, or after a spontaneous displacement of the centering sleeve of the repeater glass bushing or of the coarse scale, following vibrations and the shaking of the ship's hull by nearby explosions. The paper presents the derivations of the quantitative theoretical error estimates for various causes and applies the various expressions for 1) the determination of the magnitude and sign of the displacement of the center of the direction fiducial relative to the Card 1/2

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ACC NR: AP6001834

center of the azimuthal circle and its removal; and 2) the determination of the magnitude and sign of the displacement of the center of the coarse reading reel relative to the center of the azimuthal circle and the ways for its removal. Orig. art. has: 7 formulas, 5 figures, and 2 tables.

SUB CODE: 17 / SUBM DATE: none

H W  
Card 2/2

FESENKO, N. V.

191T63

USSR/Hydrology - Siphons

Sep 51

"Mud Filling of Siphons and Its Prevention," N. V. Fesenko, Engr

"Gidrotekh i Meliorat" Vol III, No 9, pp 60-64

Fesenko experimented in Moscow Hydraulics Inst imeni Vilyams under guidance of Professors M. V. Potapov and B. A. Pyshkin, in 1946 and 1947, studying structure of flow with directing mud guards in attempt to improve construction for achievement of max flow at min energy losses. Suggestion by P. V. Mikheyev, Cand Tech Sci, of a helical pipe is worth notice. Further study of problem is expected.

191T63

FESENKO, N. V., Cand of Bio Sci -- (diss) "Influence of light conditions on the development of generative organs of wheat grains." Leningrad, 1957, 22 pp (All-Union Academy of Agricultural Sciences im Lenin. All-Union Institute of Plant Culture), 100 copies (KL, 35-57, 107)

~~RESPETO N.V.~~, kand.biol.nauk

Physiological method of castrating wheat. Agrobiologiya no.5:90-93  
S-O '58.  
(MIRA 11:11)

1. Orlovskaya gosudarstvennaya sel'skokhozyaystvennaya opytnaya  
stantsiya.

(Wheat breeding)

FESENKO, N.V., kand. biolog. nauk

Effect of reproduction conditions on the biological and economic  
indices of buckwheat. Agrobiologija no.1:149-151 Ja-F '64  
(MIRA 17:8)

1. Orlovskaya sel'skokhozyaystvennaya opytnaya stantsiya.

FESENKO, H.Ya. (Zaporozh'ye, p.o. 9, Uchenicheskij pereulok, d. 3, kv. 4)

Female genital organs in the hernial sac of an adult male. Nov.  
khir.arkh. no.2:72-73 Mr-Ap '57. (MLRA 10:8)

1. Khirurgicheskoye otdeleniye 4-y Zaporozhskoy gorodskoy bol'nitsy  
(HEMAPHRODITISM)